## SHORT COMMUNICATION

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# Effects of storage on the binding of carboplatin to plasma proteins

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**Abstract** Plasma ultrafiltrates are routinely used in pharmacokinetic studies of carboplatin. Experiments were performed to detect and quantitate artifactual decreases in the platinum concentration of ultrafiltrates prepared from plasma samples stored at -20 °C and -70 °C. Carboplatin was added to anticoagulated, whole human blood to produce a 20 µg/ml concentration. Plasma produced from the blood was stored frozen at either -20 °C or -70 °C. Aliquots from each storage condition were thawed and ultrafiltered once a week for up to 100 days. Platinum concentrations in ultrafiltrates and plasma were determined by flameless atomic absorption spectrometry. There was no loss of ultrafilterable platinum in plasma samples stored at -70 °C, whereas there was a steady decrease in free platinum concentration in ultrafiltrates prepared from plasma samples stored at -20 °C. These results imply that pharmacokinetic studies of carboplatin should use ultrafiltrates prepared immediately or that plasma for such studies should be stored at -70 °C. Storage of carboplatincontaining plasma at -20 °C and subsequent ultrafiltration is not acceptable, because measurement of platinum in such ultrafiltrates will be artifactually low.

Key words Carboplatin · CBDCA · Protein binding

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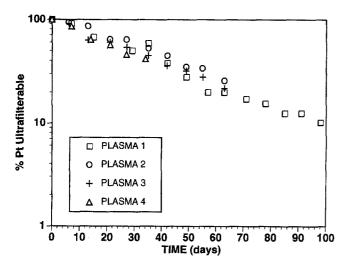
## introduction

The important platinum-containing agents, cisplatin and carboplatin, have been and continue to be frequent subjects of pharmacokinetic studies. Because both drugs are inactivated after covalent binding to plasma and other proteins, plasma ultrafiltrates are the common matrix in which concentrations of active drug are assessed [3, 6, 8, 9, 11-16, 19-21]. The known high reactivity of cisplatin with plasma proteins has resulted in the practice of immediate plasma ultrafiltration after blood samples are obtained for pharmacokinetic analysis. In contrast, carboplatin is much less reactive with plasma proteins [6, 9, 11, 15, 16, 19-21], and as a result, questions have been raised as to whether carboplatin-containing plasma could be stored frozen until it is ultrafiltered. This latter option has obvious advantages for institutions where manpower and equipment for pharmacokinetic sampling and preparation are limited. In a recent collaborative effort in which plasma samples from patients treated with carboplatin were stored at -20 °C for varying periods of time before being ultrafiltered, we found substantially less platinum in the ultrafiltrates than would be expected. As a result, we undertook a study in which we added carboplatin to whole human blood, prepared plasma from that blood, and stored the plasma at -20 °C or -70 °C to determine whether, and if so at what rate, ultrafilterable platinum is lost due to ex vivo binding during storage.

#### Materials and methods

Materials

Whole human blood from four separate individuals was collected into heparinized containers. Clinically formulated carboplatin (Paraplatin) was purchased from Bristol-Myers Squibb (Princeton, N.J.).



**Fig. 1** Concentrations of platinum in ultrafiltrates of four plasmas stored at  $-20\,^{\circ}\mathrm{C}$ 

## Methods

Within 30 min after blood was obtained, carboplatin was added to the blood to produce a final drug concentration of 20 µg/ml. Immediately thereafter, the blood was centrifuged at 1000 g for 10 min. The resulting plasma was separated into two equal volumes and frozen in 3-ml aliquots at either -20 °C or -70 °C. Thereafter, on a weekly basis, an aliquot of plasma stored at each temperature was thawed at 37 °C until liquid in consistency and immediately ultrafiltered. Ultrafiltrates were prepared by placing a portion of that plasma into Amicon Centrifree micropartition devices (Amicon Division, W.R. Grace, Beverly, Mass.) and centrifuging the devices at 2000 g for 20 min at 4 °C in a Sorvall RC-2B centrifuge (DuPont, Wilmington, Del.). Platinum concentration in plasma and ultrafiltrates was assessed with a Perkin-Elmer model 1100 flameless atomic absorption spectrometer (Perkin-Elmer, Norwalk, Conn.) monitoring 265.9 nm. The temperature program used for both plasma and ultrafiltrates was as follows: ramp over 30 s to 90 °C and hold for 60 s; ramp over 10 s to 110 °C and hold for 10 s; ramp over 30 s to 300 °C and hold for 30 s; ramp over 60 s to 1500 °C and hold for 90 s; and atomize at 2700 °C with no ramping. Argon gas flow was 300 ml/min during all heating steps except atomization when it was interrupted. Platinum concentrations were determined by comparison with a standard curve performed on the same day as the assay. When visual assessment of a semilogarithmic plot of ultrafilterable platinum vs storage time for each individual plasma stored at -20 °C indicated first-order decay, the rate of decay of ultrafilterable platinum in stored samples was determined from the formula  $A = A_0e^{-kt}$ , where A represented the amount of platinum present at any given time, t, and Ao represented the concentration of platinum present at the start of the experiment. Data were fitted using non-linear, least-squares regression as implemented by the computer program ADAPT II [2]. The half-life  $(t_{1/2})$  of ultrafilterable platinum in stored samples was calculated from the relationship:  $t_{1/2} = 0.693/k$ .

## Results

When plasma was stored at -70 °C for as long as 100 days, there was no loss of ultrafilterable platinum. In contrast, storage of plasma at -20 °C resulted in a progressive decrease in ultrafilterable platinum as storage time increased (Fig. 1). On visual inspection, this decay in ultra-

filterable platinum appeared to be a first-order process. When modelled as such, values of 0.0230, 0.0204, 0.0240, and 0.0274 were calculated for the decay constants of ultrafilterable platinum in the four respective plasma sources studied. These data indicate that storage of plasma containing carboplatin at  $-20\,^{\circ}\mathrm{C}$  is associated with an average loss of ultrafilterable platinum of approximately 2.4%/day, and imply a half-life of 29.5  $\pm$  4.0 (mean  $\pm$  SD) days for ultrafilterable platinum when plasma is stored at  $-20\,^{\circ}\mathrm{C}$ .

#### Discussion

The platinum-containing antineoplastic agent carboplatin is important because it produces therapeutic responses in a number of tumor types [1, 22] and because its pharmacokinetics have been related to both the therapeutic and the toxic consequences of its administration [4, 5, 10, 17]. As a result, carboplatin remains the subject of a considerable number of ongoing and planned pharmacokinetic studies. In recognition of the potential impact of such pharmacokinetic studies and the need for studying relatively large numbers of patients to assess pharmacokinetic/pharmacodynamic interactions, clinical studies of carboplatin with companion or intrinsic pharmacokinetic components are no longer entirely the province of the relatively limited number of institutions with extensive pharmacokinetic sampling, analytical and modelling resources. In an effort to facilitate pharmacokinetic studies at institutions with more limited personnel and equipment, a number of activities have been initiated. Among these are the development of limited sampling strategies for defining carboplatin exposure [7, 18] and the availability of designated, central analytical facilities for cooperative groups or other multi-institution, clinical trial consortia. Even with the reduced number of plasma samples involved with a limited sampling strategy, the requirement for two centrifugation steps to convert whole blood to plasma and subsequently plasma ultrafiltrate can strain the resources of many clinical centers. A simpler procedure would be to prepare plasma in the clinical setting and store it under appropriate conditions until a later time when: (1) multiple ultrafiltrates could be prepared as a batch; or (2) ultrafiltrate could be produced at the site where flameless atomic absorption spectrometry or other analysis of platinum content would be performed. These considerations were not raised previously in studies of cisplatin because no storage conditions were felt to be capable of preventing that compound's high reactivity from resulting in ex vivo binding to plasma proteins and artifactual reduction in platinum concentration of plasma ultrafiltrate. The less reactive nature of carboplatin has led to the question of whether plasma from patients treated with that drug could be stored frozen until plasma ultrafiltrates could be produced at a later time.

Our data demonstrate that storage of carboplatin-containing plasma at -20 °C is not suitable. Accurate determination of ultrafilterable platinum requires production of

plasma ultrafiltrate immediately after production of plasma or immediately after thawing of plasma stored at -70 °C. Investigators involved in planning or reviewing pharmacokinetic studies of carboplatin should be aware of these limitations.

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